CPC COOPERATIVE PATENT CLASSIFICATION

C CHEMISTRY; METALLURGY

(NOTES omitted)

CHEMISTRY

C01 INORGANIC CHEMISTRY

(NOTES omitted)

C01C AMMONIA; CYANOGEN; COMPOUNDS THEREOF ({metal hydrides, monoborane, diborane or addition complexes thereof C01B 6/00}; salts of oxyacids of halogens C01B 11/00; peroxides, salts of peroxyacids C01B 15/00; thiosulfates, dithionites, polythionates C01B 17/64;

compounds containing selenium or tellurium <u>C01B 19/00</u>; azides <u>C01B 21/08</u>; {compounds other than ammonia or cyanogen, containing nitrogen, non-metals and optionally metals <u>C01B 21/082</u>}; metal imides or amides <u>C01B 21/092</u>; nitrites <u>C01B 21/50</u>; {compounds of noble gases <u>C01B 23/0005</u>}; phosphides <u>C01B 25/08</u>; salts of oxyacids of phosphorus <u>C01B 25/16</u>; compounds containing silicon <u>C01B 33/00</u>; compounds containing boron <u>C01B 35/00</u>)

WARNING

In this subclass non-limiting references (in the sense of paragraph 39 of the Guide to the IPC) may still be displayed in the scheme

1/00	Ammonia; Compounds thereof {(C01C 3/08, C01C 3/14, C01C 3/16, C01C 3/20 take precedence)}	1/0458	• • • • {Separation of NH ₃ (during purge gas treatment <u>C01C 1/0476</u>)}
	NOTE	1/0464	• • • • {by absorption in liquids, e.g. water}
		1/047	• • • • {by condensation}
	Complex ammine salts, e.g. [Pd(NH3)4]Cl2, are { also} classified in the relevant groups of	1/0476	 {Purge gas treatment, e.g. for removal of inert gases or recovery of H₂}
	subclasses <u>C01D</u> - <u>C01G</u> , according to the metal	1/0482	• • • {Process control; Start-up or cooling-down procedures}
1/003	• {Storage or handling of ammonia}	1/0488	* * * * * * * * * * * * * * * * * * * *
1/006	 {making use of solid ammonia storage materials, e.g. complex ammine salts} 	1/0400	• • • {Processes integrated with preparations of other compounds, e.g. methanol, urea or with processes for power generation}
1/02	 Preparation, {purification} or separation of 	1/0404	
	ammonia	1/0494	• • {using plasma or electric discharge}
1/022	 {Preparation of aqueous ammonia solutions, i.e. ammonia water} 	1/08	• • Preparation of ammonia from nitrogenous organic substances
1/024	• • {Purification}	1/083	• • • (from molasses (treatment of molasses in
1/026	• • {Preparation of ammonia from inorganic		general <u>C13B 50/006</u>)}
	compounds}	1/086	{from urea}
1/028	• • • {from ammonium sulfate or sulfite}	1/10	Separation of ammonia from ammonia liquors,
1/04	• Preparation of ammonia by synthesis {in the gas		e.g. gas liquors {(as part of the ammonia
	phase}(preparation or purification of gas mixtures		synthesis process <u>C01C 1/04</u>)}
	for ammonia synthesis {C01B 3/025})	1/12	Separation of ammonia from gases and vapours
1/0405	$\{from N_2 and H_2 in presence of a catalyst\}$		{(as part of the ammonia synthesis process C01C 1/04)}
1/0411	• • • {characterised by the catalyst}	1/14	Saturators
1/0417	• • • {characterised by the synthesis reactor,	1/14	Halides of ammonium
	e.g. arrangement of catalyst beds and heat		
	exchangers in the reactor (arrangement of	1/162	• • {Ammonium fluoride}
	several reactors <u>C01C 1/0405</u> ; fixed-bed	1/164	• {Ammonium chloride}
	reactors in general <u>B01J 8/02</u>)}	1/166	• • {Ammonium bromide}
1/0423	{Cold wall reactors}	1/168	{Ammonium iodide}
1/0429	• • • • {Fluidized or moving bed reactors}	1/18	. Nitrates of ammonium
1/0435	{Horizontal reactors}	1/185	• • {Preparation}
1/0441	• • • • {Reactors with the catalyst arranged in	1/20	• Sulfides; Polysulfides
	tubes}	1/22	Sulfites of ammonium
1/0447	• • • {Apparatus other than synthesis reactors}	1/24	• Sulfates of ammonium (<u>C01C 1/14</u> takes
1/0452	{Heat exchangers}		precedence)

CPC - 2025.08

1/242	• Preparation from ammonia and sulfuric acid or sulfur trioxide	3/0283	• • • {from simple or complex cyanides of the noble metals}
1/244	Preparation by double decomposition of ammonium salts with sulfates	3/0287	• • • {from simple or complex cyanides of other transition metals, e.g. from iron cyanides}
1/245	Preparation from compounds containing nitrogen and sulfur	3/0291	• • • {from simple or complex cyanides of other metals}
1/246	from sulfur-containing ammonium compounds	3/0295	• • {Purification}
1/247	by oxidation with free oxygen	3/04	Separation from gases
1/248	Preventing coalescing or controlling form or size	3/06	Stabilisation of hydrogen cyanide
	of the crystals	3/08	 Simple or complex cyanides of metals
1/249	 Deacidifying {or drying} the crystals 		{(<u>C01C 3/001</u> , <u>C01C 3/002</u> take precedence)}
1/26	 Carbonates or bicarbonates of ammonium 	3/10	Simple alkali metal cyanides
1/28	. Methods of preparing ammonium salts in general	3/11	Complex cyanides
1,20		3/12	Simple or complex iron cyanides
	<u>NOTES</u>		
	1. This group does not cover ammonium salts of	3/14	• Cyanic {or isocyanic} acid; Salts thereof
	complex acids (other than complex cyanides)		$\{(\underline{\text{C01C 3/001}} \text{ takes precedence})\}$
	containing a metal in the anion, which are	3/145	• • {Isocyanic acid; Salts thereof}
	covered by the relevant groups of subclasses	3/16	• Cyanamide; Salts thereof ({C01C 3/001,
			<u>C01C 3/002</u> takes precedence}; dicyandiamide
	C01D - C01G, according to the metal.		<u>C07C 279/28</u>)
	2. Salts of polybasic acids with ammonium and	3/18	Calcium cyanamide
	a metal as cations are classified as though the	3/20	• Thiocyanic acid; Salts thereof {(C01C 3/001 takes
	ammonium were hydrogen.	2,20	precedence)}
3/00	Cyonogon, Compounds thousaf		procedure,)
	Cyanogen; Compounds thereof		
3/001	• {Preparation by decomposing nitrogen-containing		
	organic compounds, e.g. molasse waste or urea (by		
	distillation of carbamates <u>C01C 3/02</u> , <u>C01C 3/08</u> ,		
	C01C 3/14, C01C 3/16; by decomposing formamide		
	or ammonium formate <u>C01C 3/0204</u>)}		
3/002	• {Synthesis of metal cyanides or metal cyanamides		
	from elementary nitrogen and carbides}		
3/003	• {Cyanogen}		
3/004	• {Halogenides of cyanogen}		
3/005	• {Thiocyanogen}		
3/006	• {Sulfurdicyanide}		
3/007	• {Ammonium cyanide}		
3/008	• {Cyanazide}		
3/02	 Preparation, {separation or purification} of 		
	hydrogen cyanide {(C01C 3/001 takes precedence)}		
3/0204	• • {from formamide or from ammonium formate}		
3/0208	• • {Preparation in gaseous phase}		
3/0212	• • • {from hydrocarbons and ammonia in the		
3/0212	presence of oxygen, e.g. the Andrussow-		
	process}		
2/0216	*		
3/0216	• • • {characterised by the catalyst used}		
3/022	{Apparatus therefor}		
3/0225	• • • • {characterised by the synthesis reactor}		
3/0229	• • • {from hydrocarbons and ammonia in the		
	absence of oxygen, e.g. HMA-process}		
3/0233	• • • • {making use of fluidised beds, e.g. the		
	Shawinigan-process}		
3/0237	• • • {from carbon monoxide and ammonia}		
3/0241	• • • {from alcohols or aldehydes}		
3/0245	• • {from organic nitriles, e.g. acetonitrile}		
3/0243			
	• • {by using a plasma}		
3/0254	• • {from cyanates or from thiocyanates}		
3/0258	• • {from cyanamides or derivatives thereof}		
3/0262	• • {from cyanides}		
3/0266	{from simple alkali or alkaline earth metal		
	cyanides}		
3/027	{Alkali metal cyanides}		
3/027	{Alkaline earth metal cyanides}		
3/0279	• • • {from ammonium cyanide}		

CPC - 2025.08